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answer to question #49691, chemistry, inorganic chemistry - answer to question #49691, chemistry, inorganic chemistry what's the sodium hydroxide solution is standardized using 0.502 g of solid acid $H_2C_2O_4$. find the molarity of the base if 30.50 ml are required for the titration to a phenolphthalein endpoint? solution: $H_2C_2O_4$

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ap chemistry scoring guidelines, 2016 - college board - a student investigates the enthalpy of solution, ΔH_{soln} , for two alkali metal halides, $LiCl$ and $NaCl$. in addition to the salts, the student has access to a calorimeter, a balance with a precision of ± 0.1 g, and a

practice test ch 3 stoichiometry name per - remember it is a mc test, use the answers ... practice test ch 3 stoichiometry name _____ per _____ $2MnO_2 + 4KOH + O_2 + Cl_2 \rightarrow 2KMnO_4 + 2KCl + 2H_2O$. for the reaction above, there is 100. g of each reactant available. which reagent is the limiting reagent? [molar masses: MnO_2

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